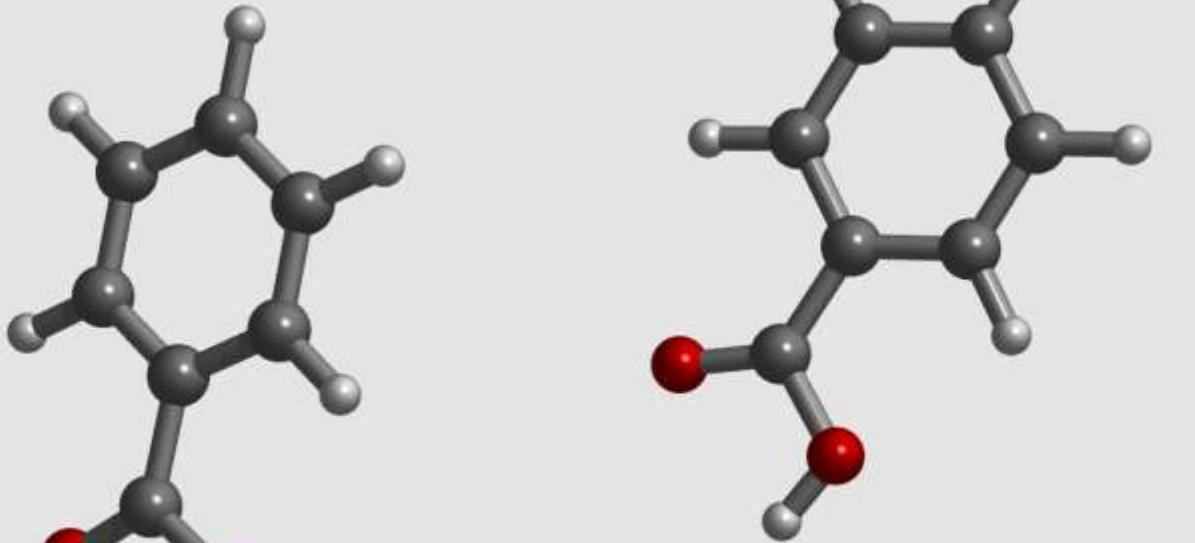
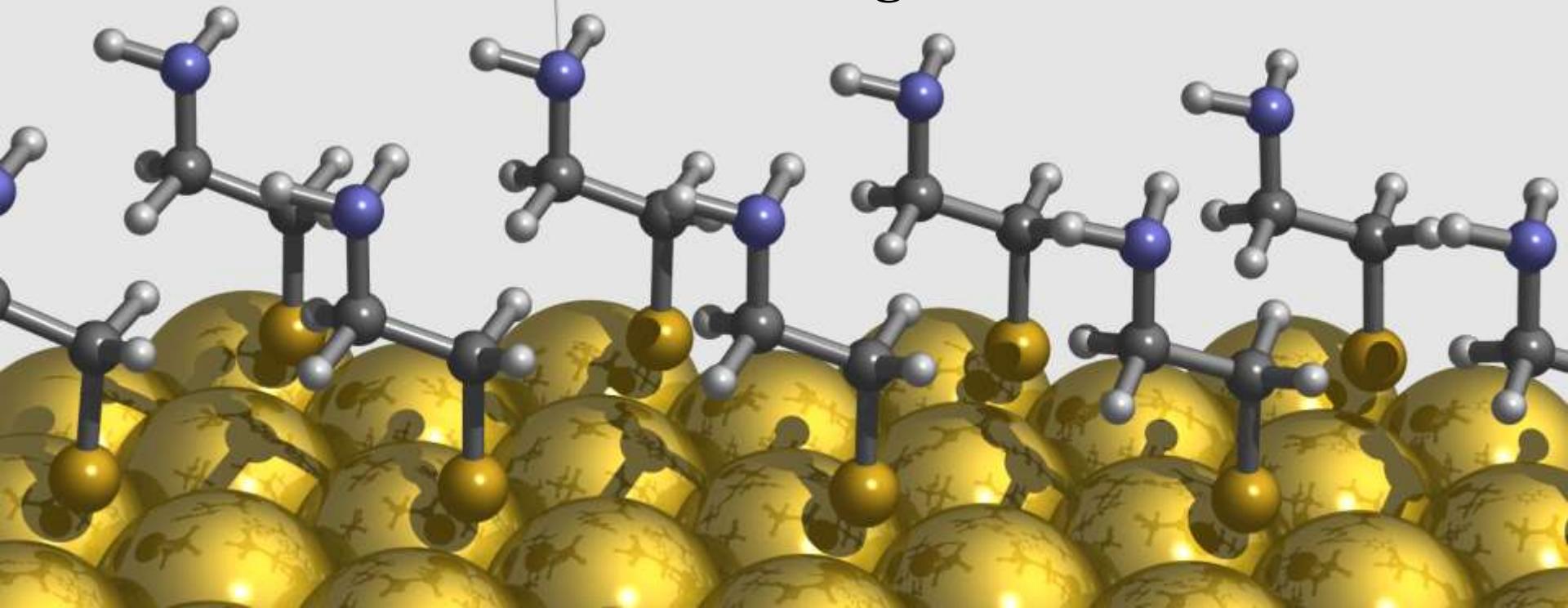


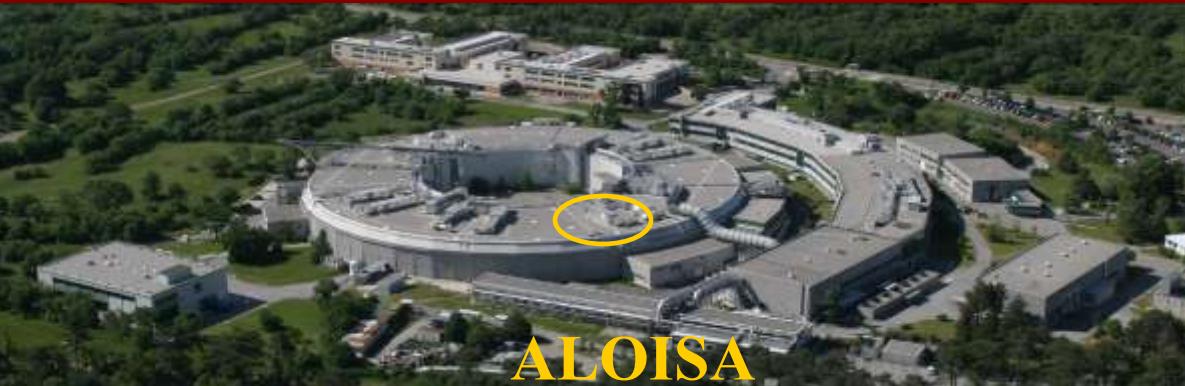
A. Cossaro
CNR-IOM
Trieste



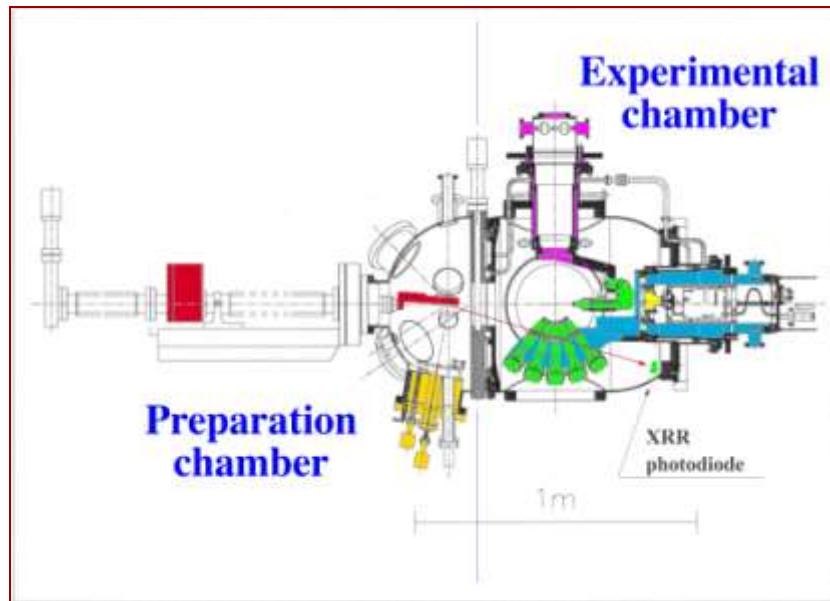
Tailoring SAM on SAM formation



ALOISA beamline



Elettra, Trieste



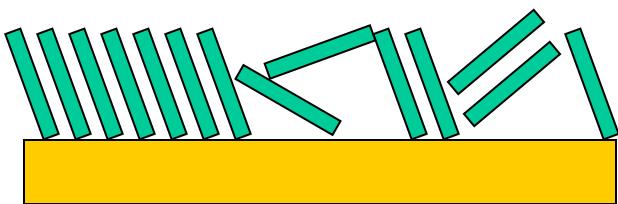
Luca Floreano
Albano Cossaro
Alberto Verdini
CNR-IOM

Alberto Morgante
CNR-IOM
Università di Trieste

H_v energy range: 120-8000 eV
High Resolution Photoemission Spectroscopy
X-Ray Photoelectron Diffraction
Grazing Incidence X-Ray Diffraction
X-Ray Reflectivity
Absorption Spectroscopy
Photoemission Coincidence Spectroscopy

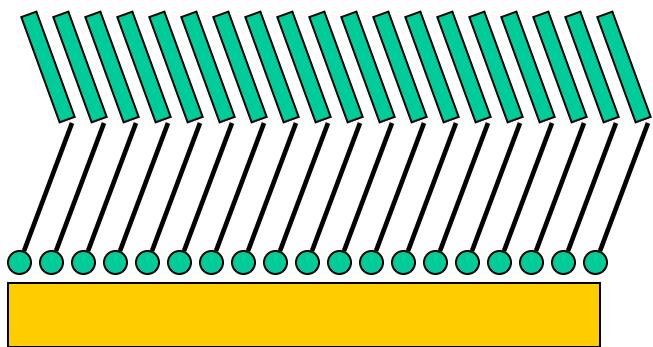
Dean Cvetko
Gregor Bavdek
Gregor Kladnik
Ljubljana University

Metal-organic interfaces

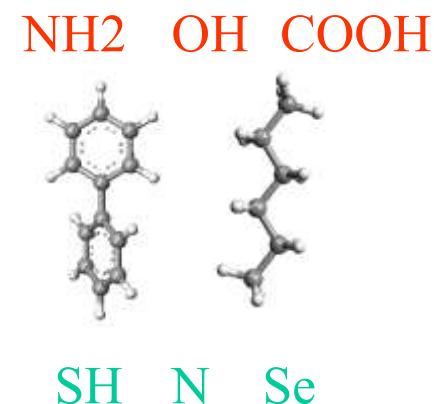
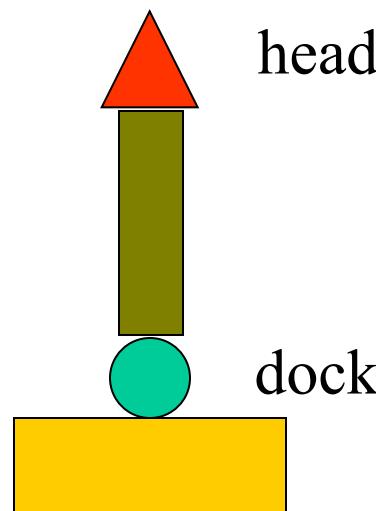


Electronic transport efficiency
System reliability

SAM interposition



Homogeneous interface morphology
Improved electronic transport



SH N Se

Tailoring of the Work Function
Electronic levels alignment

Our approach

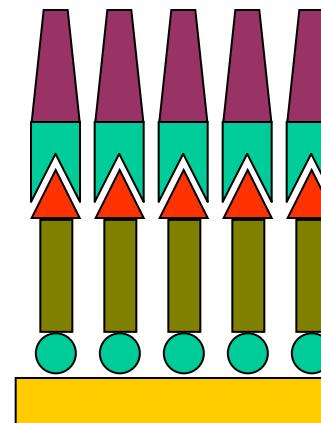
film-film



UHV

functionalization

molecule-molecule



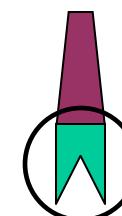
Control on the growth
In-situ X-ray spectroscopy

Modelling the junction as
sum of sub-units

ω -alkanethiols on Au(111)



Dense packing
Thermal stability
Homogeneity



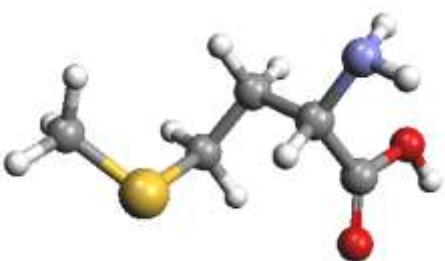
COOH



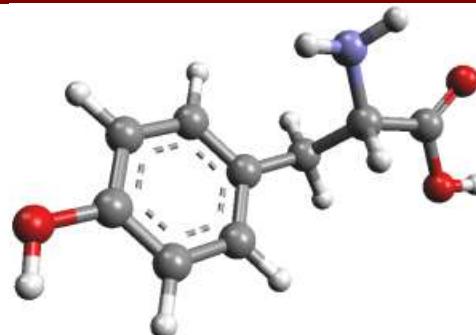
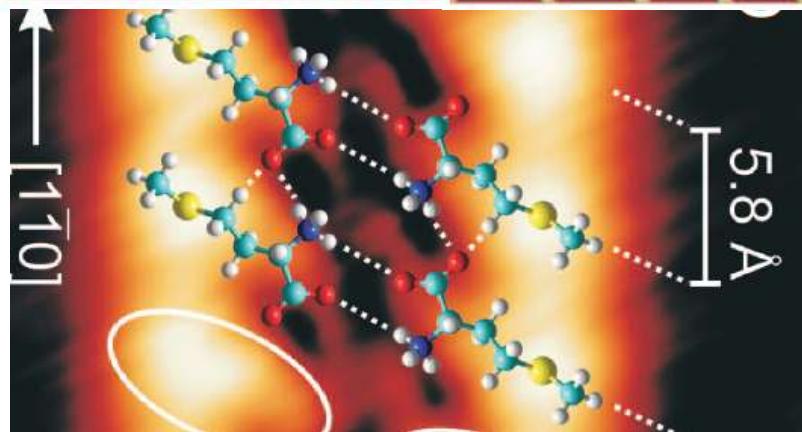
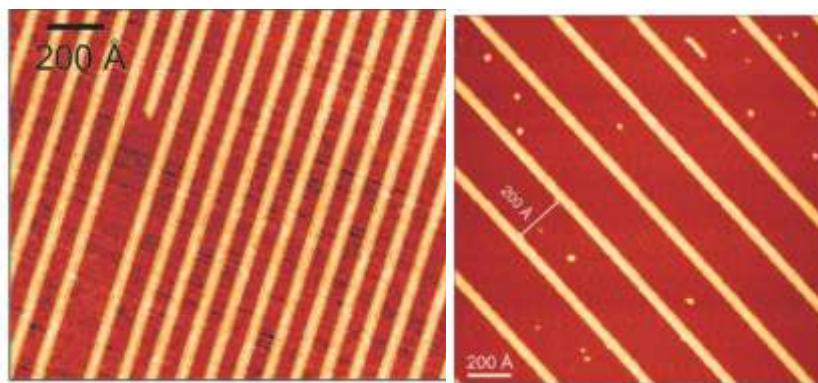
NH₂

aminoacids self-assembly

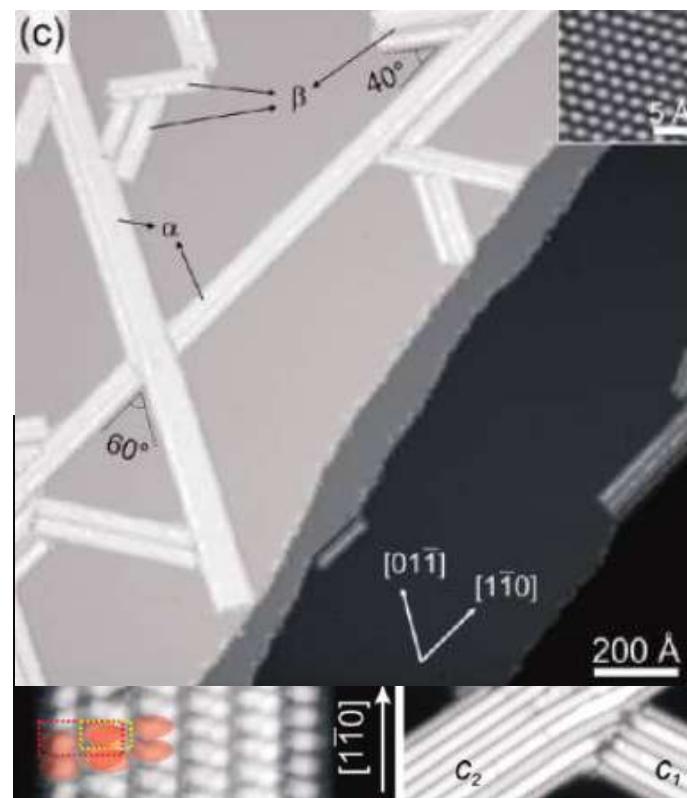
The NH₂-COOH chemical affinity



L-methionine
on Ag(111)



L-tyrosine
on Ag(111)



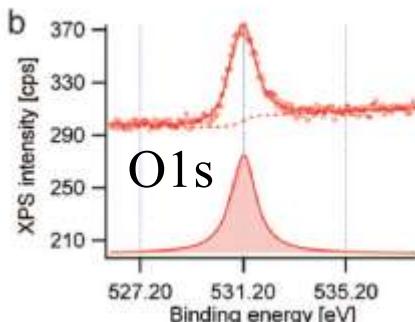
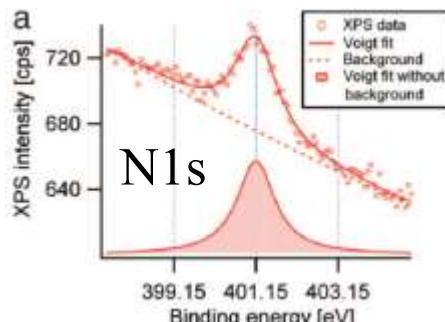
in coll. with J. Barth

PNAS 2007

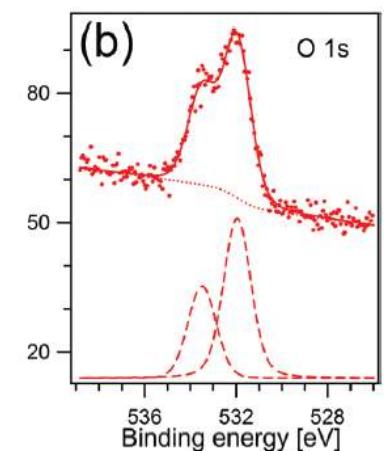
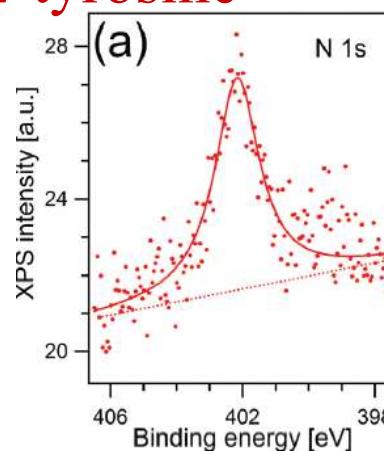
ACS Nano 2010

The NH₂-COOH chemical affinity

L-methionine



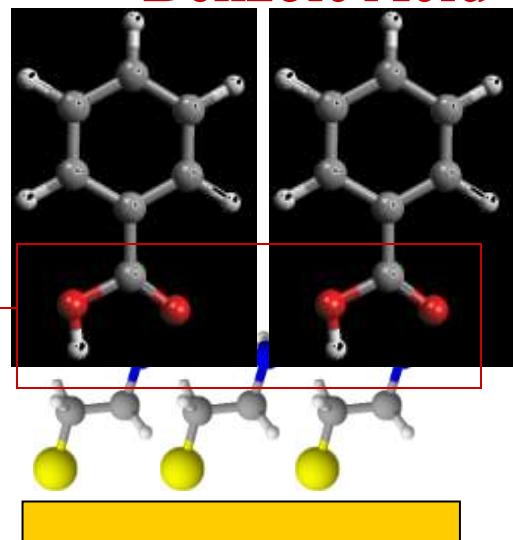
L-tyrosine



molecules in their zwitterionic form

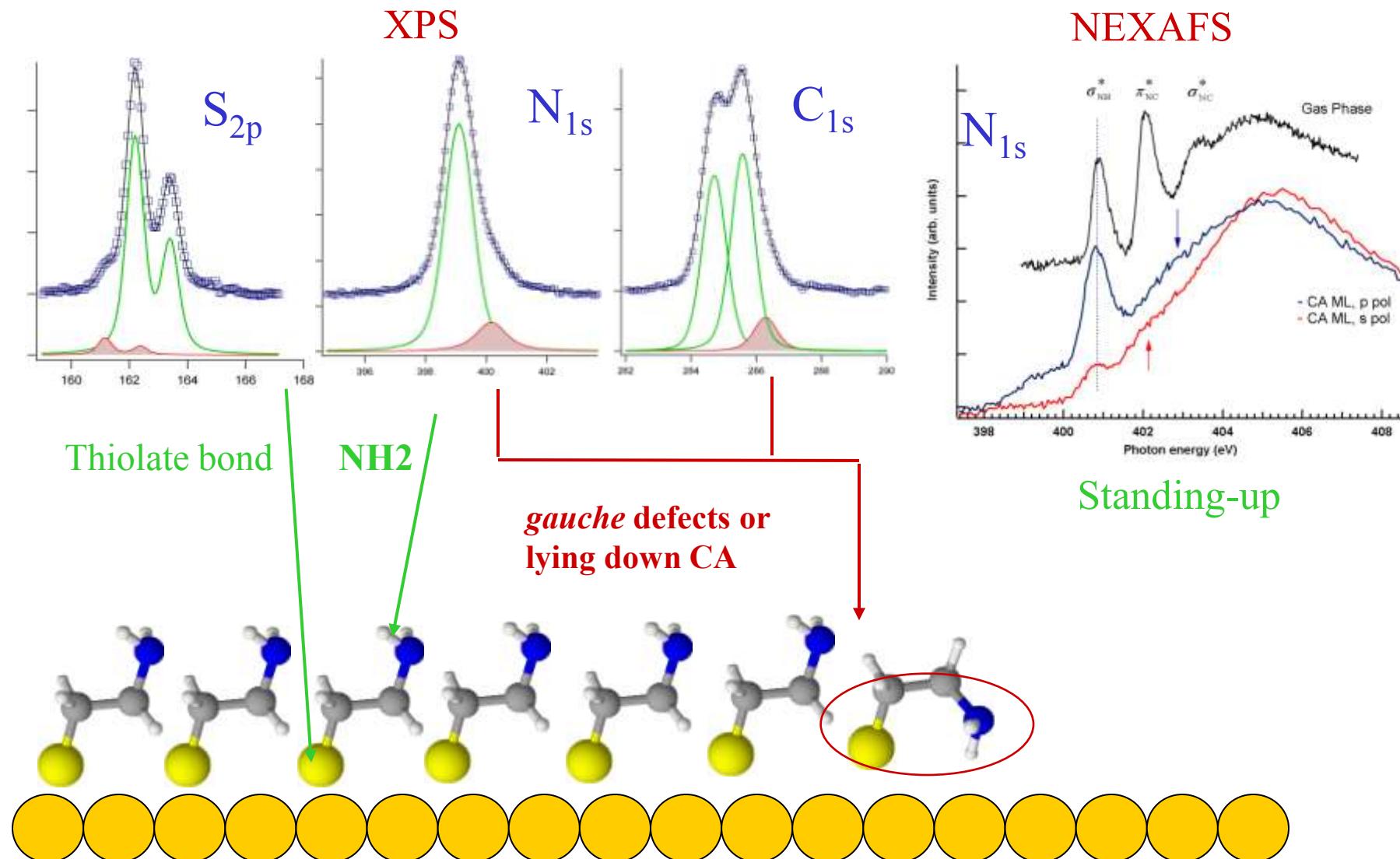
COO⁻-NH₃⁺ interaction drives the formation of templates of biomolecules

Benzoic Acid

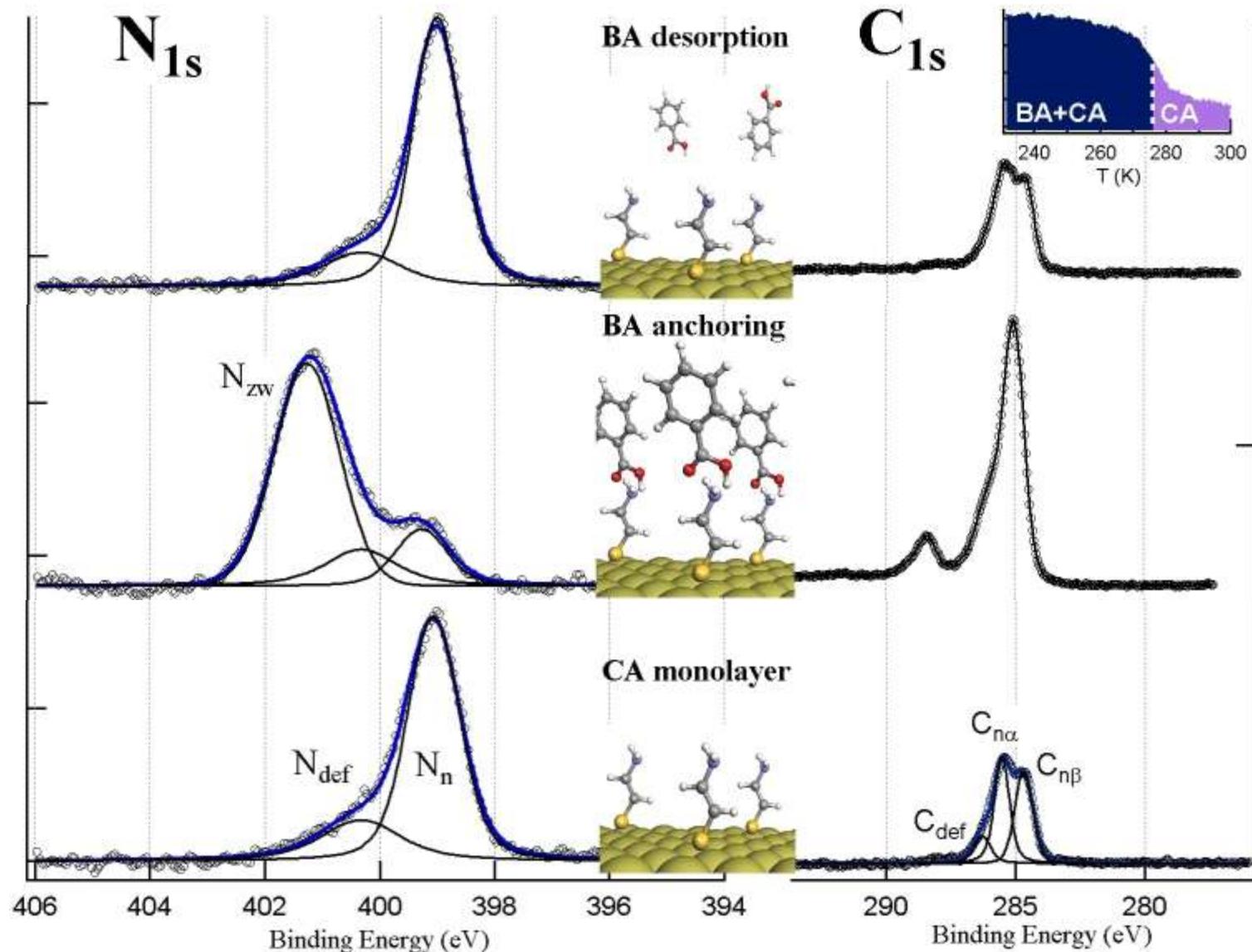


Cysteamine / Au(111)

1. Functionalization : in-situ CA monolayer formation



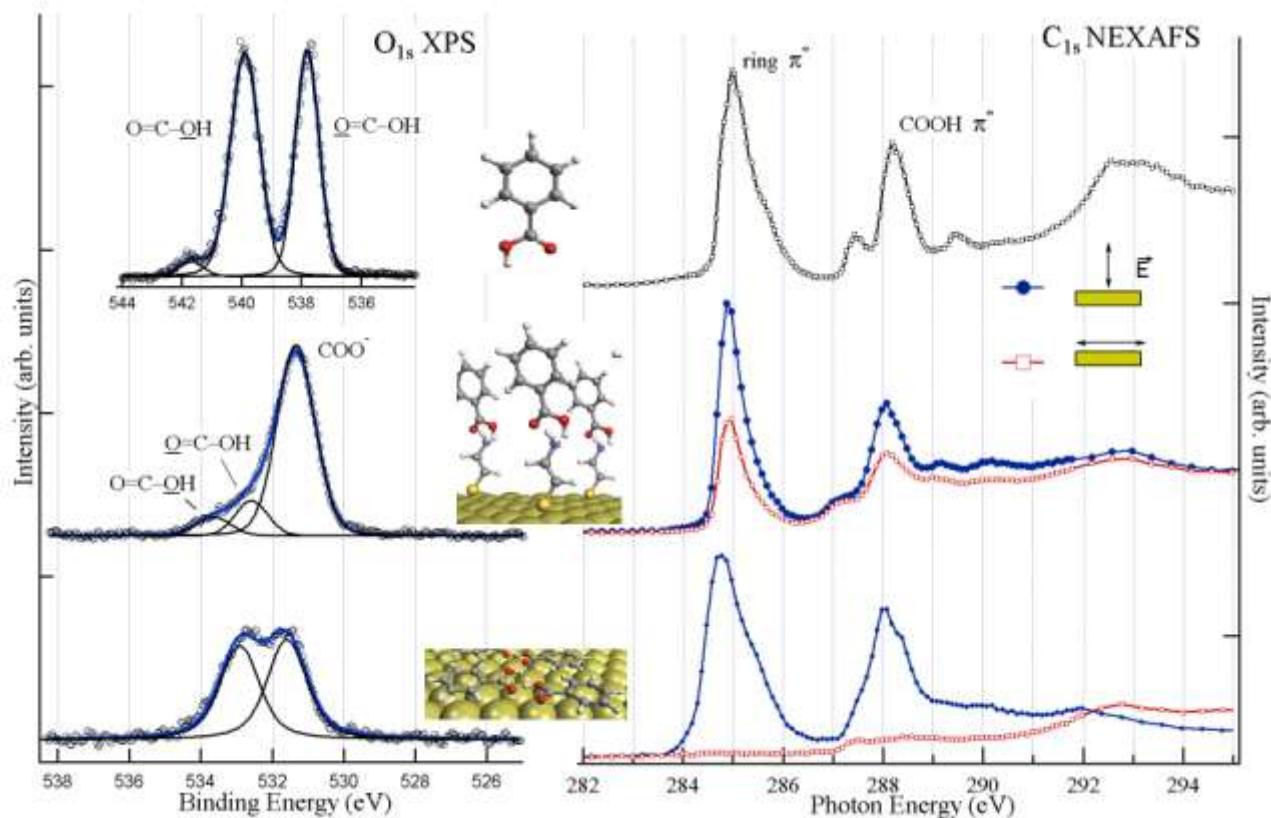
2. Anchoring: Benzoic Acid Deposition on CA



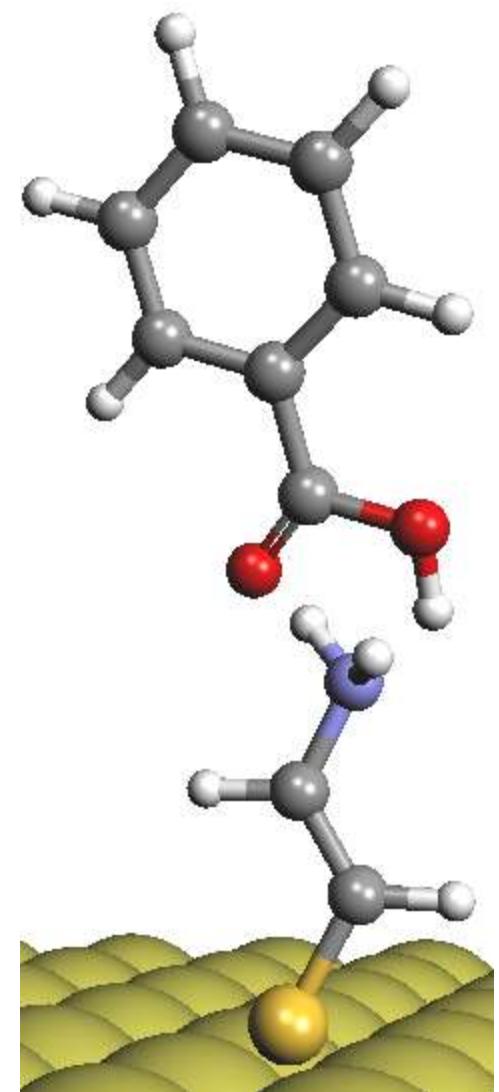
NH₃⁺-COO⁻ bond formation
High anchoring efficiency

J. Phys. Chem. Lett. 2011

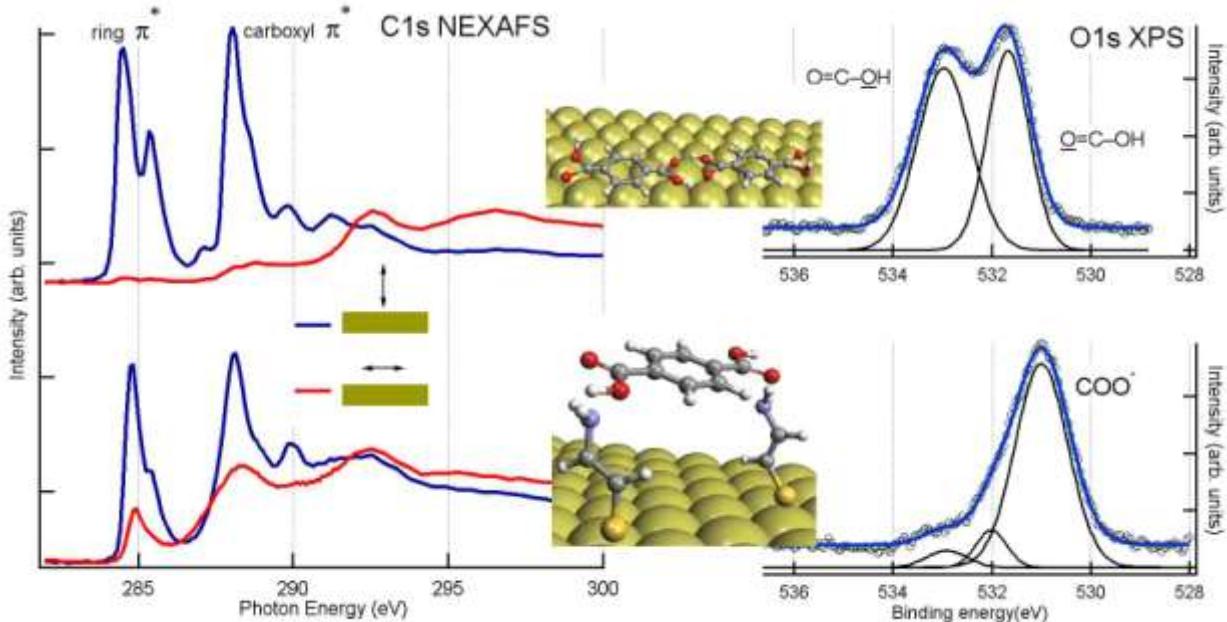
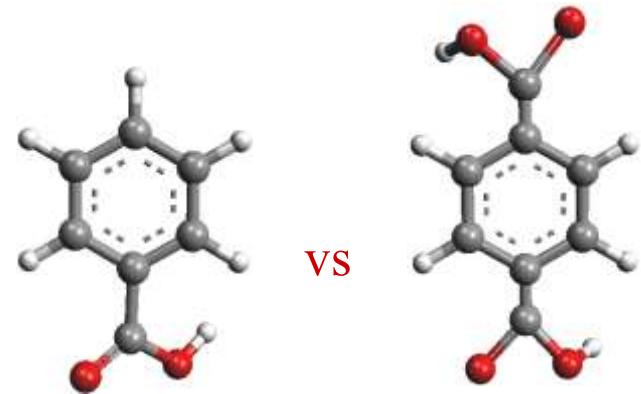
2. Anchoring: Benzoic Acid Deposition on CA



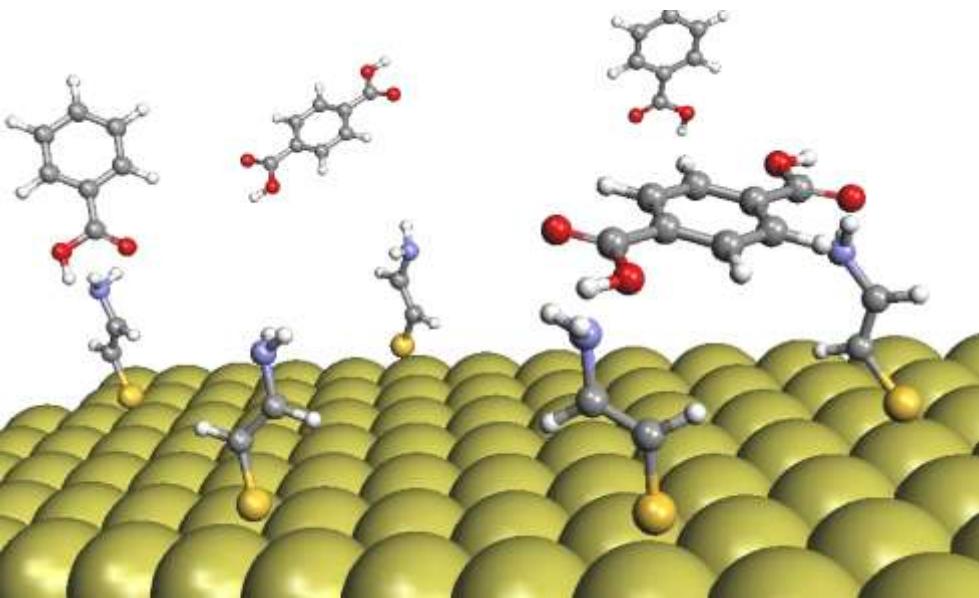
most of BA are anchored
anchoring changes the adsorption geometry



Bi-dentate anchoring: TPA vs BA



TPA: lying down adsorption
bi-dentate anchoring



Morphology changes by selecting
the number of anchoring groups

Spectroscopy

Resonant Photoemission

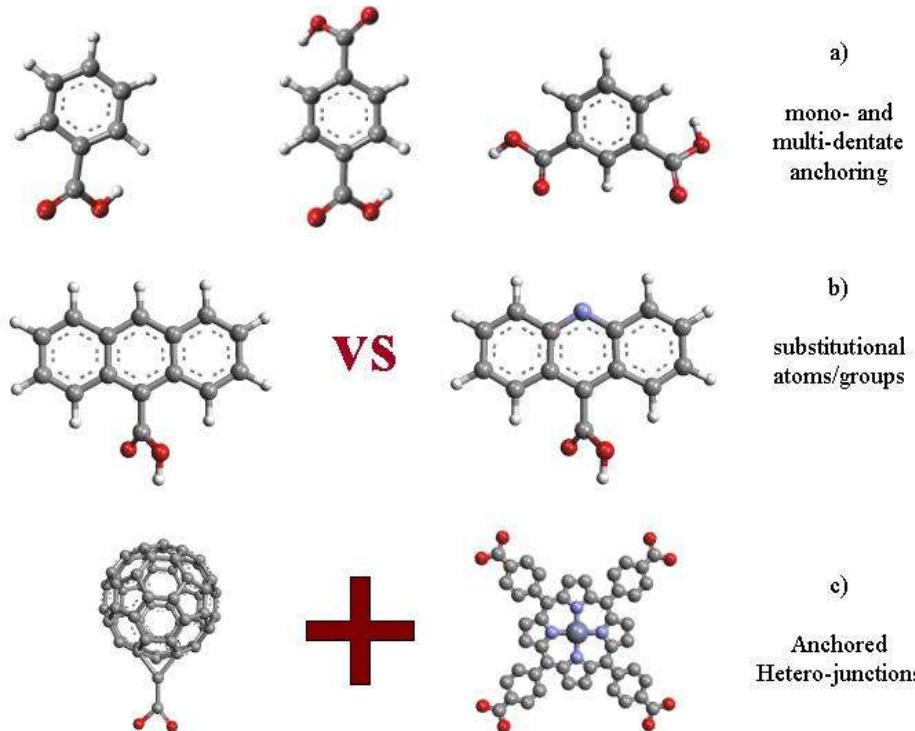
DFT

Amino-carboxylic bond

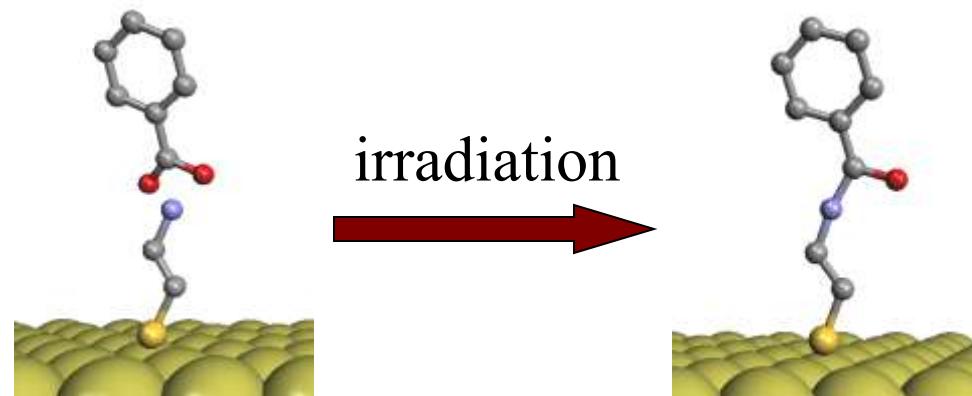
STM Imaging

STS - conductance

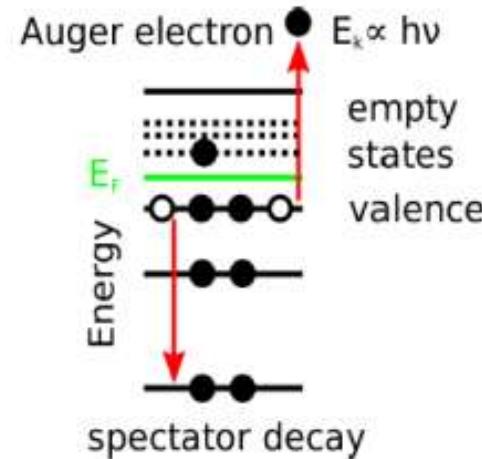
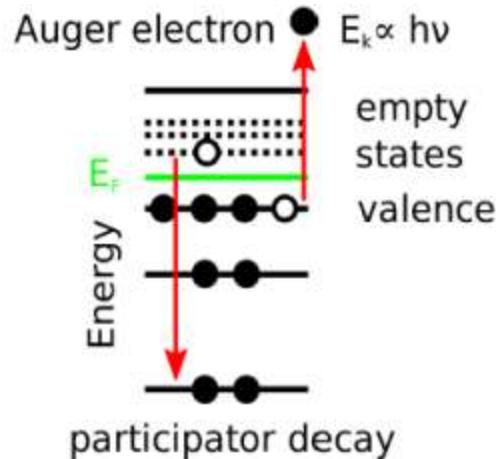
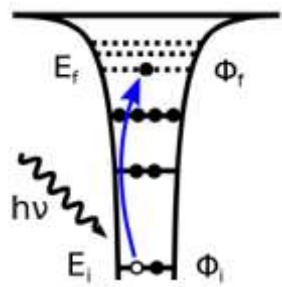
Carlo Dri, Università di Trieste



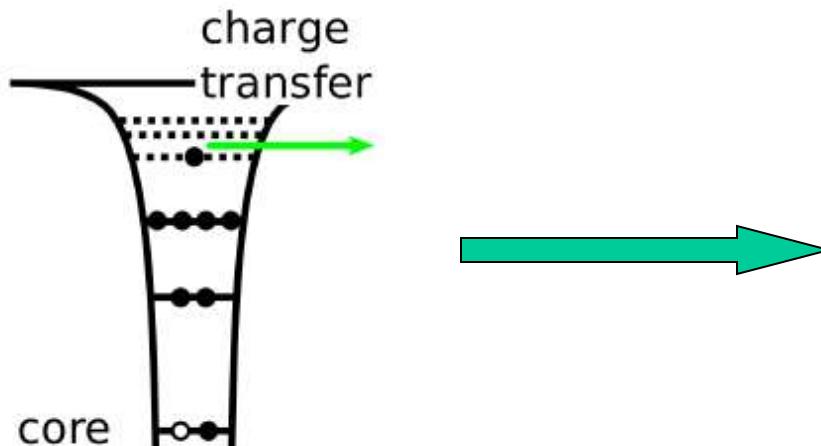
Covalent bond formation



Resonant X-Ray Photoemission



1. Resonant excitation

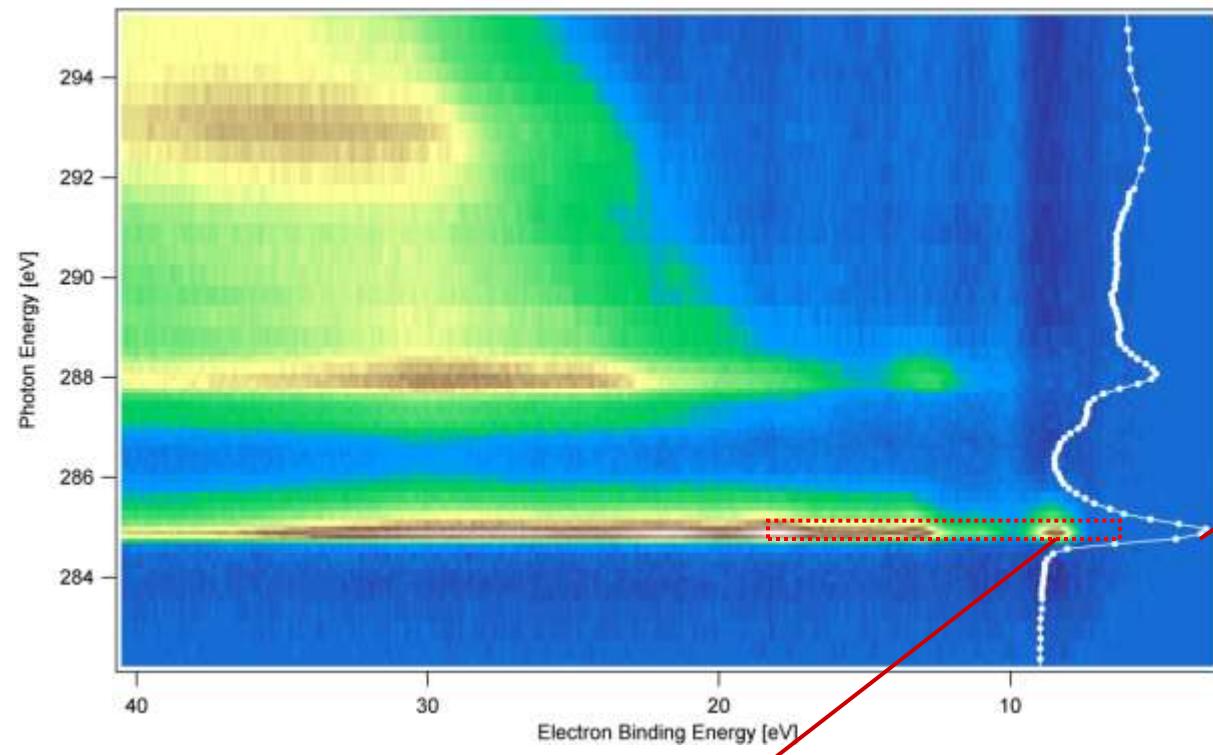


2. non radiative decay channels

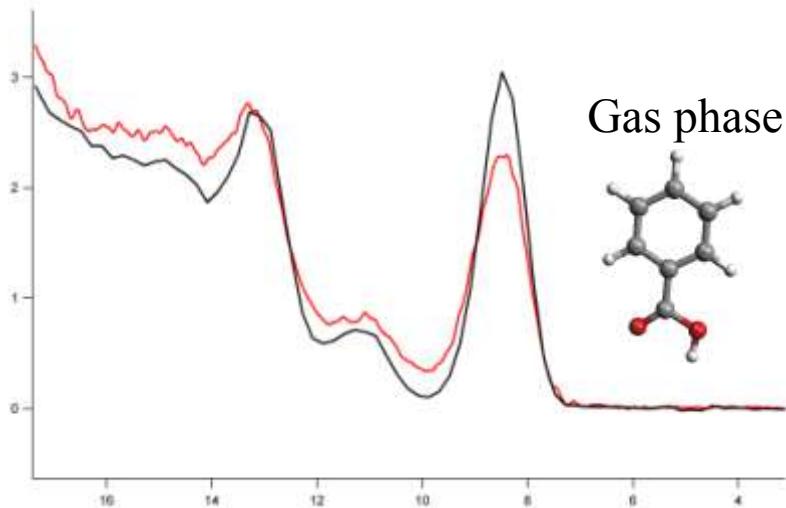
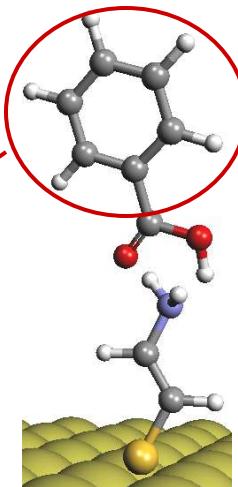
Normal Auger Decay

Quenching of the resonance
in the Valence Band

Resonant Photoemission: first results



C1s edge



Gas phase

Quenching of the resonance
due to the anchoring process

Evidence of BA/Au electronic transport
mediated by CA

ALOISA

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MIUR Futuro in Ricerca

Prot. RBFR10FQBL

